

## 3-(4-Chlorophenyl)-3-hydroxy-1-phenylpropan-1-one

Zhi-Fang Yu,\* Xiu-Yan Gu, Bing Zhao and Zhong-Zhen Tian

Department of Chemistry, Tianjin University,  
Tianjin 300072, People's Republic of ChinaCorrespondence e-mail:  
zhifang@public.tpt.tj.cn

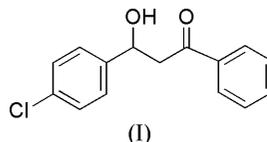
## Key indicators

Single-crystal X-ray study  
 $T = 293$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.005$  Å  
 $R$  factor = 0.045  
 $wR$  factor = 0.167  
Data-to-parameter ratio = 13.8For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

The title compound,  $\text{C}_{15}\text{H}_{13}\text{ClO}_2$ , was synthesized by a Reformatsky reaction in an aqueous medium. The two benzene rings are approximately parallel. Molecules are connected by intermolecular  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds, forming extended chains in the  $a$  axis direction.

## Comment

We have recently investigated the Reformatsky reaction (Bieber *et al.*, 1997) in aqueous media (Chan *et al.*, 1994; Shen *et al.*, 1997). A new compound, namely 3-(4-chlorophenyl)-3-hydroxy-1-phenylpropan-1-one, (I), has been synthesized by the reaction of 4-chlorobenzaldehyde and 2-bromoacetophenone in the presence of zinc in an aqueous medium. An X-ray crystal structure determination of (I) was carried out in order to elucidate the structure and the results are presented here.



The molecular structure of (I) is illustrated in Fig. 1. The two benzene rings are approximately parallel, forming a dihedral angle of  $5.3(2)^\circ$ , different from what is found in 3-(2,4-dichlorophenyl)-3-hydroxy-1-phenylpropan-1-one (Yu *et al.*, 2003), where the two benzene rings are orthogonal. The angle  $\text{C}10-\text{C}9-\text{C}8$  is  $121.0(2)^\circ$ , indicating that  $\text{C}9$  is  $sp^2$  hybridized. Molecules related by unit-cell translations are connected by intermolecular  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds (Table 2), forming extended chains in the  $a$ -axis direction.

## Experimental

The title compound was synthesized by adding a mixture of 4-chlorobenzaldehyde (0.421 g, 3 mmol) and 2-bromoacetophenone (0.892 g, 4.5 mmol) to a mixture of a saturated solution of calcium chloride (12 ml), zinc (0.781 g, 12 mmol), ammonium chloride (1.5 g), a trace amount of iodine, cetyltrimethylammonium bromide (0.005 g) and THF (1 ml). The mixture was stirred at room temperature for 2 h. and extracted with diethyl ether, evaporated and separated by flash chromatography (eluent ethyl acetate–petroleum ether). A colourless powder was obtained (yield 82%) and single crystals suitable for crystallographic analysis were obtained by slow evaporation of an ethyl acetate–petroleum ether solution (m.p. 370–371 K). Spectroscopic analysis: IR (KBr,  $\nu$ ,  $\text{cm}^{-1}$ ): 3468, 1668;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ,  $\delta$ , p.p.m.): 7.95–7.24 (*m*, 9H), 5.30 (*t*, 1H), 3.32 (*m*, 2H). Analysis required for  $\text{C}_{15}\text{H}_{13}\text{ClO}_2$ : C 69.10, H 4.99%; found: C 69.21, H 4.91%.

Received 15 December 2004  
Accepted 20 December 2004  
Online 8 January 2005

Crystal data

C<sub>15</sub>H<sub>13</sub>ClO<sub>2</sub>  
*M<sub>r</sub>* = 260.70  
 Monoclinic, *P*<sub>2</sub><sub>1</sub>/*c*  
*a* = 5.360 (9) Å  
*b* = 8.697 (14) Å  
*c* = 27.46 (5) Å  
 β = 90.17 (3)°  
*V* = 1280 (4) Å<sup>3</sup>  
*Z* = 4

*D<sub>x</sub>* = 1.353 Mg m<sup>-3</sup>  
 Mo Kα radiation  
 Cell parameters from 1001 reflections  
 θ = 3.0–25.1°  
 μ = 0.29 mm<sup>-1</sup>  
*T* = 293 (2) K  
 Prism, colorless  
 0.30 × 0.20 × 0.06 mm

Data collection

Bruker SMART CCD area-detector diffractometer  
 φ and ω scans  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
*T<sub>min</sub>* = 0.933, *T<sub>max</sub>* = 0.983  
 6347 measured reflections

2270 independent reflections  
 1511 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.038  
 θ<sub>max</sub> = 25.0°  
*h* = -4 → 6  
*k* = -10 → 10  
*l* = -29 → 32

Refinement

Refinement on *F*<sup>2</sup>  
*R* [*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.045  
*wR* (*F*<sup>2</sup>) = 0.167  
*S* = 1.07  
 2270 reflections  
 164 parameters

H-atom parameters constrained  
*w* = 1/[σ<sup>2</sup>(*F<sub>o</sub>*<sup>2</sup>) + (0.0966*P*)<sup>2</sup>]  
 where *P* = (*F<sub>o</sub>*<sup>2</sup> + 2*F<sub>c</sub>*<sup>2</sup>)/3  
 (Δ/σ)<sub>max</sub> < 0.001  
 Δρ<sub>max</sub> = 0.24 e Å<sup>-3</sup>  
 Δρ<sub>min</sub> = -0.19 e Å<sup>-3</sup>

Table 1

Selected geometric parameters (Å, °).

O1–C7	1.428 (3)	O2–C9	1.223 (3)
O2–C9–C10	120.0 (2)	C13–C14–C15	119.9 (3)
C10–C9–C8	121.0 (2)		
O1–C7–C8–C9	66.4 (3)	O2–C9–C10–C15	177.7 (2)
C6–C7–C8–C9	-167.9 (2)		

Table 2

Hydrogen-bonding geometry (Å, °).

<i>D</i> –H... <i>A</i>	<i>D</i> –H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> –H... <i>A</i>
O1–H1...O2 <sup>1</sup>	0.82	2.032	2.847 (4)	172.35

Symmetry code: (i) 1 + *x*, *y*, *z*.

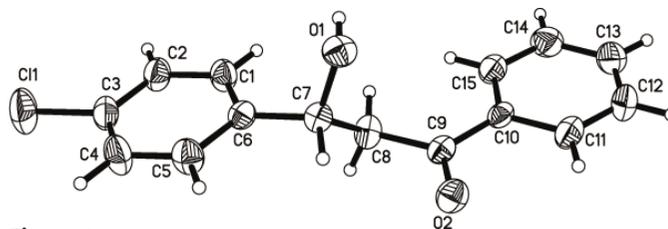


Figure 1

View of the molecule of (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 30% probability level.

All H atoms were positioned geometrically and refined as riding (O–H = 0.82 Å and C–H = 0.93–0.98 Å), with *U<sub>iso</sub>*(H) values of 1.2 (CH and CH<sub>2</sub>) or 1.5 (OH) times *U<sub>eq</sub>* of the parent atom.

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

The authors thank the State Key Laboratory of Elemento-Organic Chemistry, Nankai University. This research was supported by the Visiting Scholar Foundation of Key Labs in the university. The authors thank Tianjin University. The work was also supported by '985' Project of Tianjin University.

References

- Bieber, L. W., Malvestiti, I. & Storch, E. C. (1997). *J. Org. Chem.* **62**, 9061–9064.  
 Bruker (1997). SMART, SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Chan, T. H., Li, C. J. & Lee, M. C. (1994). *Can. J. Chem.* **72**, 1181–1192.  
 Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.  
 Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.  
 Shen, Z., Zhang, J. Q., Zou, H. X. & Yan, M. M. (1997). *Tetrahedron Lett.* **38**, 2733–2736.  
 Yu, Z. F., Zhao, B., Tian, Zh. & Gu, X. Y. (2003). *Acta Cryst.* **E59**, o2020–o2021.